

ARTICLE

Received 9 Aug 2013 | Accepted 2 Jun 2014 | Published 15 Jul 2014

DOI: 10.1038/ncomms5275

Atomic scale real-space mapping of holes
in $\text{YBa}_2\text{Cu}_3\text{O}_{6+\delta}$ N. Gauquelin^{1,†}, D.G. Hawthorn², G.A. Sawatzky³, R.X. Liang³, D.A. Bonn³, W.N. Hardy³ & G.A. Botton¹

The high-temperature superconductor $\text{YBa}_2\text{Cu}_3\text{O}_{6+\delta}$ consists of two main structural units—a bilayer of CuO_2 planes that are central to superconductivity and a $\text{CuO}_{2+\delta}$ chain layer. Although the functional role of the planes and chains has long been established, most probes integrate over both, which makes it difficult to distinguish the contribution of each. Here we use electron energy loss spectroscopy to directly resolve the plane and chain contributions to the electronic structure in $\text{YBa}_2\text{Cu}_3\text{O}_6$ and $\text{YBa}_2\text{Cu}_3\text{O}_7$. We directly probe the charge transfer of holes from the chains to the planes as a function of oxygen content, and show that the change in orbital occupation of Cu is large in the chain layer but modest in CuO_2 planes, with holes in the planes doped primarily into the O 2p states. These results provide direct insight into the local electronic structure and charge transfers in this important high-temperature superconductor.

¹Department of Materials Science and Engineering, Brockhouse Institute and Canadian Centre for Electron Microscopy, McMaster University, 1280 Main Street West, Hamilton, Ontario, Canada L8S 4M1. ²Department of Physics & Astronomy, University of Waterloo, 200 University Avenue West, Waterloo, Ontario, Canada N2L 3G1. ³Canadian Institute for Advanced Research, Canada, and Department of Physics & Astronomy, University of British Columbia, 6224 Agricultural Road, Vancouver, British Columbia, Canada V6T 1Z1. † Present address: EMAT, University of Antwerp, Groenenborgerlaan 171, 2020 Antwerp, Belgium. Correspondence and requests for materials should be addressed to G.A.B. (email: gbotton@mcmaster.ca).

YBa₂Cu₃O_{6+δ} (YBCO) has been extensively studied since its discovery, as it was the first material known to be superconducting above liquid nitrogen temperature. Although electron energy loss spectroscopy (EELS)^{1–6} and X-ray absorption spectroscopy (XAS)^{2,7–12} of YBCO compounds have been studied over the full range of oxygen stoichiometry, the presence of multiple, distinct copper and oxygen sites (see crystal structure in Fig. 1a), each having different orbital symmetries and doping dependencies, give rise to complex spectra at the Cu *L* and O *K* edge. In these studies, the contributions to the spectra from different sites could not be

spatially resolved, requiring reliance on the symmetry of the orbital occupation and their doping dependence to infer contributions to the spectra from different lattice sites. However, the conclusions drawn from these symmetry and doping arguments retain some ambiguity because the different Cu and O sites do not generally have unoccupied states with pure, well-defined symmetries (pure $d_{x^2-y^2}$ versus $d_{y^2-z^2}$) that can be neatly resolved as a function of photon polarization. Moreover, due to the strongly correlated nature of the cuprates, standard *ab initio* theoretical tools such as density functional theory capture some important details of electronic structure but have proven

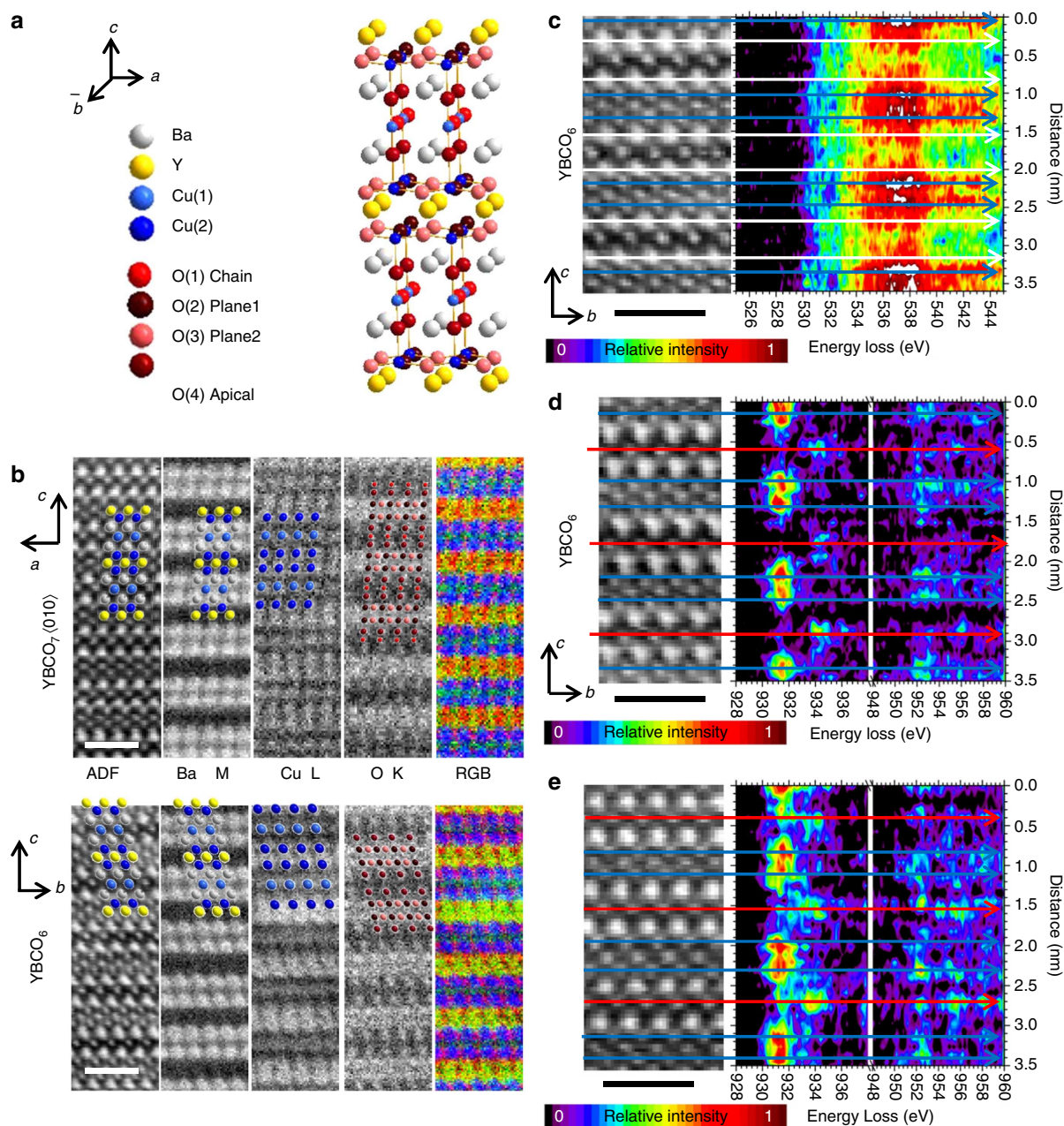


Figure 1 | Atomically resolved elemental maps and 2D representation of the O *K* and Cu *L*_{2,3} edges in YBCO. (a) The crystal structure of YBa₂Cu₃O₇ showing multiple, distinct atomic sites. (b) Atomic EELS elemental maps of YBCO₇ [010] (top) and YBCO₆ [100] (bottom) showing PCA-treated (10 components) maps of the Ba M_{4,5}, Cu L_{2,3} and O *K* edges. The coloured image is an RGB combination of Ba (blue), O (red) and Cu (green) maps. Spatially resolved EELS for (c) YBCO₆ at the O *K* edge, (d) YBCO₆ at the Cu *L* edge and (e) YBCO₇ at the Cu *L* edge. These spectra are integrated over the (*ab*) atomic planes but spatially resolved along the *c* axis. The red, blue and white lines indicate the atomic planes corresponding to the chain layer (Cu₁), CuO₂ plane layer (Cu₂, O_{2,3}) and BaO planes (O₄), respectively. Clear differences between the plane and chain layers are observed in the spectra. The scale bars are 1 nm.

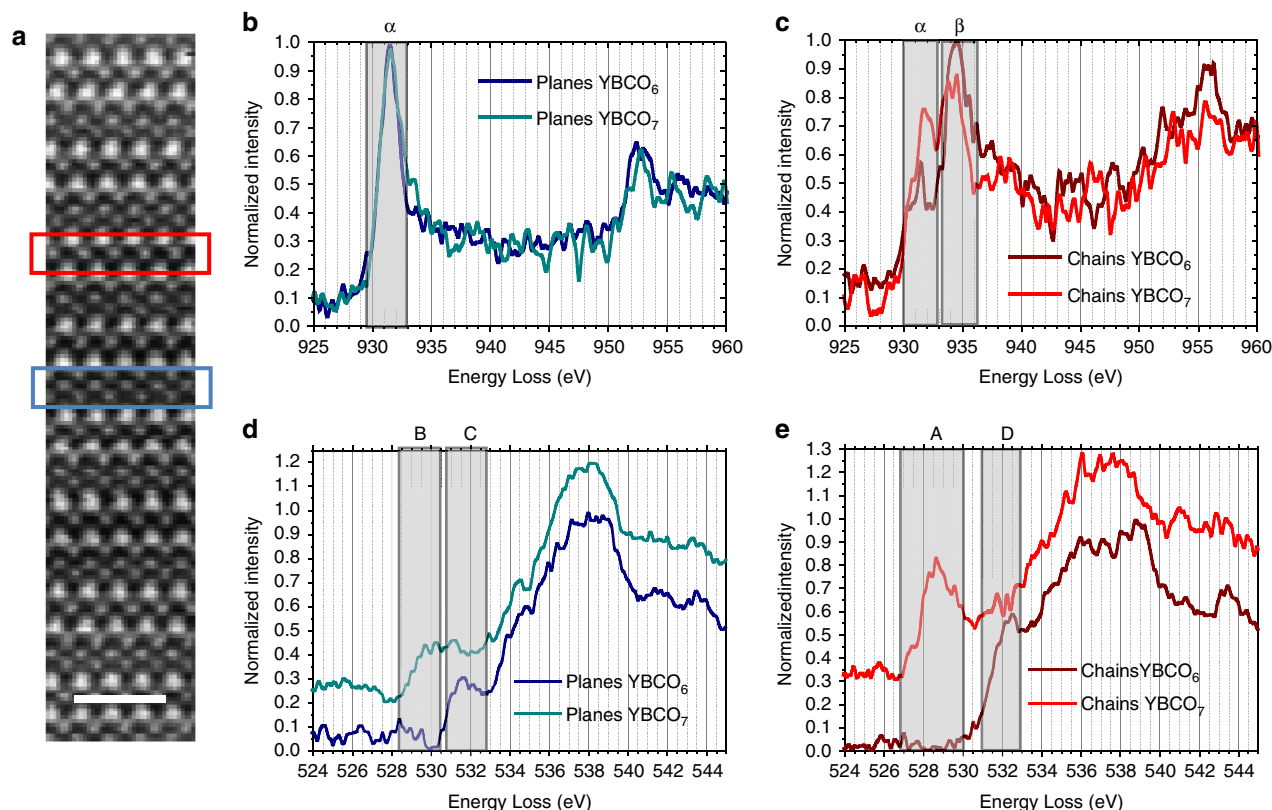


Figure 2 | Energy-loss near-edge structure of chains and planes in doped and undoped YBCO. (a) High-angle annular dark-field image showing the definition of the regions used for the chains and planes spectra (red rectangle and blue rectangle, respectively). Comparison of the raw Cu L (b,c) and O K (d,e) between doped and undoped YBCO in planes (b,d) and chains (c,e) showing differences between the different electronic structures. Electronic transitions are marked as described in the text. Energy scales were aligned to data reported by XAS by Fink *et al.*² Scale bar, 1 nm.

unreliable in predicting details of the cuprate electronic structure. Accordingly, a more complete description of the electronic structure of YBCO requires more direct means to distinguish the electronic structure of different sites.

In this report, we make use of recent developments in EELS resolution^{13–16} to measure O K and Cu L_{2,3} spectra in YBa₂Cu₃O_{6+δ} with sufficient spatial and energy resolution to clearly distinguish the role of the planes and chains in real space. We show atomically resolved elemental maps for both doped and undoped YBCO as well as fine structure differences between chains and planes at both the Cu L and the O K edges. We demonstrate that, on doping YBCO from O₆ to O₇, the atomically resolved data of the Cu L edge spectra in the planes remain very similar, whereas at the O K edge a lower energy peak arises corresponding to doped holes running through E_F (consistent with the Zhang–Rice singlet band) and justifying the description of the cuprates as doped charge transfer insulators. Our results are contrasted with recent spatially resolved EELS measurements of YBCO:LSMO multilayers¹⁷, which exhibit a different O K edge energy dependence.

In the chain layer, changes of spectral weight at the Cu L edge indicate that the additional dopant O transforms Cu from Cu¹⁺ to Cu²⁺. At the O K edge, a large transfer of spectral weight occurs as the doping is changed from O₆ to O₇, from a high energy peak corresponding to empty chain states to a lower energy peak corresponding to full chain charge carrier states. The spatially resolved results are compared with polarized XAS data and shown to be largely consistent with the orbital symmetry assignments of the different Cu and O sites found in previous studies. Finally, we examine a novel means to explore the doping-dependent electronic structure of YBCO by modifying the O content through progressive electron beam exposure.

Results

Atomically resolved elemental maps. Single crystals of YBa₂Cu₃O_{6+δ} with two different oxygen stoichiometries (δ = 0.03 and 0.97, hereafter referred to as YBCO₆ and YBCO₇) were measured. Figure 1b shows atomically resolved maps of YBCO₇ (top panel) taken in the <010> direction and YBCO₆ (bottom panel) with the corresponding scanning transmission electron microscopy (STEM) annular dark-field (ADF) image. These maps show that the chemical nature of the atomic sites in this complex structure can be readily identified and that the experimental mapping conditions are suitable to identify the edges of interest. Further maps are presented on Supplementary Fig. 1. The oxygen atomic columns are clearly visible on the O K edge map. We note, however, that the oxygen (O(1)) atoms in the chains are not observed clearly in YBCO₇, whether in the [100] or [010] directions. For the [010] projection, this might be linked to the presence of Cu in the same atomic column, whereas in the [100] projection, half the amount of oxygen present in the BaO and CuO₂ planes is present in the column, which might hinder the visibility. Furthermore, the contrast in oxygen atomic maps is strongly influenced by the neighbouring heavy atoms¹⁸.

Doping dependent EELS in chains and planes. Figure 1c presents the spatially resolved O K edge EELS spectra integrated over horizontal lines and plotted in a two-dimensional (2D) distribution in which the horizontal axis is the energy loss, the vertical axis is the position in the unit cell along the c axis, and the intensity is shown as a colour scale. The use of this representation to display EELS spectra provides a simple visualization of changes in the fine structure, layer by layer. The atomic

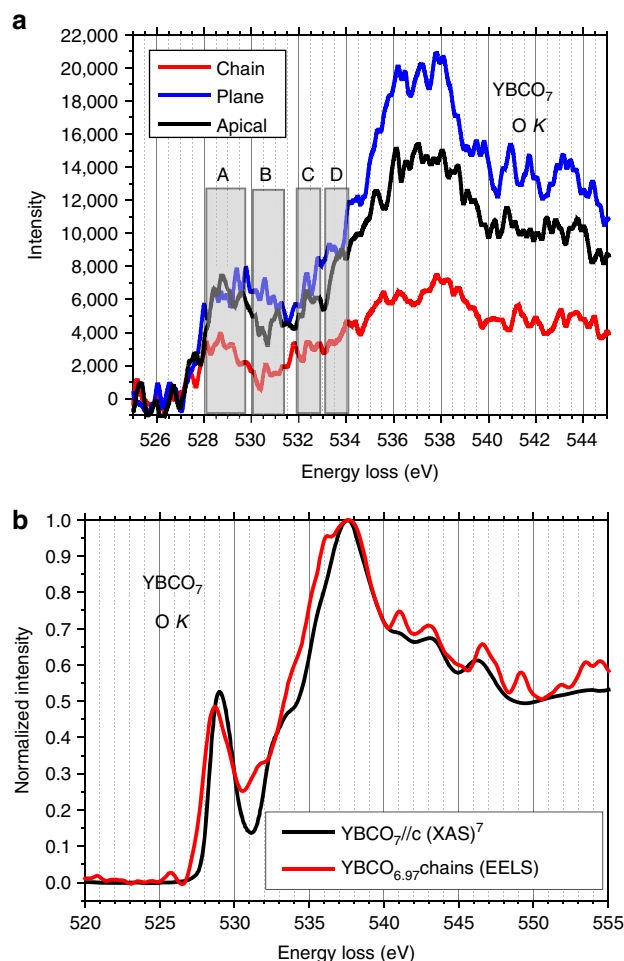


Figure 3 | EELS site-specific O K edge spectra and agreement with XAS. (a) Site-specific raw O K edge spectra of YBCO₇ per unit cell. (b) Comparison of XAS from ref. 7 and EELS using a convolution with a Gaussian distribution corresponding to the resolution of the EELS measurement.

resolution of our experiment in the *c* direction is confirmed, as differences can be seen between the plane and chain in this measurement geometry. This result will be discussed in detail further below by comparing the detailed spectra resolved in the different sites. On a similar 2D projection of the Cu L_{2,3} edge of both YBCO₆ (Fig. 1d) and YBCO₇ (Fig. 1e), we can clearly resolve significant differences between the Cu L spectra in the chains and in the planes between doped and undoped YBCO, with further details visible in the detailed spectra shown in Fig. 2.

Electronic structure differences at the Cu L edge. To investigate the difference in the electronic structure between the planes and chains, the EELS spectra from the planes and chains are extracted from the spectrum images by summing pixels over the related area (see Fig. 2a). To reduce the effect of beam exposure (discussed below), the spectra have been selected from the first two unit cells of the spectrum image in the case of YBCO₇. In YBCO₆, which is more stable, the signal could be averaged on identical atomic sites over 10 unit cells since no beam exposure dependence was seen. The Cu L₃ and L₂ white lines arise from transitions of $2p_{3/2} \rightarrow 3d_{3/2}3d_{5/2}$ and $2p_{1/2} \rightarrow 3d_{3/2}$, respectively. These edges are very sensitive to valence states as they are related to unoccupied states in the transition metal 3d band. The intensity of the Cu L edge from the chains is significantly lower

than the one coming from the planes, as can be seen in Supplementary Fig. 2.

In YBCO₆, the Cu L edge spectra in the plane layer (Fig. 2b) has a Lorentzian-like peak (labelled α) consistent with Cu²⁺ ($3d^9$), whereas the spectra in the chain layer is predominantly peaked at 934.5 eV (labelled β), similar to Cu¹⁺ in Cu₂O (ref. 19). The small peak around 931.5 eV can be linked to residual oxygen in the chains forming Cu²⁺. These results are generally consistent with XAS and EELS studies of YBCO, as well as with other cuprates, such as La_{2-x}Sr_xCuO₄ (LSCO), that have only a CuO₂ plane layer^{20,21}. (A comparison between polarization-dependent XAS and our spatially resolved EELS is provided in Supplementary Fig. 3.) The polarization (*q*)-dependent XAS (EELS) of spectra has been used to distinguish the planes and chains via the different orbital symmetries of the unoccupied states in the planes and chains. In contrast, the present measurements distinguish the planes and chains, and how they evolve with doping, much more directly, without any ambiguity due to the different Cu sites, and do so in real space.

Of great significance, and once more without ambiguity due to the multiple sites, we demonstrate that, on doping YBCO from O₆ to O₇, the atomically resolved data of the Cu L edge spectra in the planes remain very similar. However, the chain layer (Fig. 2c) exhibits a pronounced enhancement in intensity around 932 eV (slightly higher than the peak in the planes at 931.5 eV) and a reduction in intensity around 934.5 eV. This doping evolution is consistent with the additional electronegative O in the chain layer transforming the Cu in the chain layer from Cu¹⁺ to Cu²⁺ (with the holes in the Cu²⁺ of the chains occupying $3d_{y^2-z^2}$ symmetry states, as opposed to the $3d_{x^2-y^2}$ symmetry states of Cu²⁺ in the planes). The doped oxygen is also well known to dope additional holes into the CuO₂ planes, with O₇ roughly corresponding to $P = 0.194$ additional holes per CuO₂ plane²². However, according to previous XAS and EELS studies, these doped holes are understood to go primarily into the O 2p states rather than the Cu 3d orbitals, consistent with our measurements. Accordingly, the evolution of the Cu 3d states in the planes is modest and is identified by a $3d^9 L$ peak at ~ 932.5 eV that grows roughly linearly with doping^{7,9,20} and is evident in the present experiments as a slight increase in intensity at 933 eV in the O₇ sample relative to the undoped O₆ sample (Fig. 2b).

Electronic structure differences at the O K edge. At the O K edge, the doping evolution of the O 2p states in the planes is more pronounced than in the Cu L edge. In the plane layer (Fig. 2d), the EELS spectra of the O₆ sample has an onset of intensity at ~ 530 eV with a peak at ~ 531 eV (labelled C). When doped, a lower energy peak (labelled B) emerges corresponding to doped holes running through E_F (consistent with the Zhang–Rice singlet band). This doping dependence is in accordance with the doping-dependent XAS and spatially integrated EELS of the pre-edge in LSCO^{21,23}, the results demonstrate (1) that the doped holes locate predominantly into the O 2p states (specifically with p_x and p_y symmetry²⁰) and (2) that the cuprates are strongly correlated systems (specifically doped charge transfer insulators), as evidenced by a transfer of spectral weight from the upper Hubbard band (UHB) (identified in our measurements as peak C) to states near the Fermi energy (peak B), rather than a simple band shift as one would have in a conventional insulator. In YBCO, a similar transfer of spectral weight of the two lowest energy pre-peaks is observed for XAS that, with light polarized along the *a* axis of the sample, is sensitive to the O $2p_x$ symmetry states^{2,8,9}. The correspondence between our spatially resolved EELS of the plane layer and *E*//*a* XAS (see Supplementary Fig. 4) provides strong validation for the assignment of these *E*//*a* XAS pre-peaks to states in the planes and subsequently for the basic

description of the cuprates as doped charge transfer insulators. This agreement is notable as our EELS of the plane layer differ markedly from the EELS of Chien *et al.*¹⁷ on the YBCO planes in LSMO:YBCO multilayers, which show no discernable pre-peak structure and a broad peak at ~ 532.5 eV. The reason for these significant differences between the two studies is unclear, but they may be indicative of a difference in the basic electronic structure of YBCO in an LSMO:YBCO multilayer relative to a single crystal sample. In contrast, the O *K* spectra of the chain layer, discussed below, demonstrate a better qualitative agreement between our work and Chien *et al.*¹⁷, although the O doping of the sample in Chien *et al.*¹⁷ is not known.

In the chain layer (Fig. 2e), the O *K* edge EELS also exhibits a large transfer of spectral weight, with a peak at 532.5 eV (peak D) that diminishes and a lower energy peak at 529 eV (peak A) that emerges as the doping is changed from O₆ to O₇. These spectral features can be identified with ‘empty’ (peak D) and ‘full’ (peak A) chain sites. Peak A has a clear correspondence with polarization-dependent XAS with the polarization along the *b* or *c* axes of the sample. (Figure 3b presents a scaled and Gaussian broadened comparison between XAS//*c* from ref. 7 and EELS of the chains in YBCO₇, confirming this correspondence.) With these polarizations, XAS is sensitive to the O *p_y* and *p_z* symmetry orbitals, which are expected to be the lowest unoccupied orbitals of the O(1) and O(4) oxygen sites, respectively. Peak D, corresponding to empty chain states, however, is not very clearly identified on the basis of structural symmetry in polarization-dependent XAS. Here we show unambiguously that these states are identified with the chain layer and, more specifically, with the O(1) site. We can therefore infer from this new data that, on doping, supplementary holes in the planes, corresponding to the Zhang–Rice singlet, are going to states with oxygen character, whereas in the chains, hole charge carriers are formed with mixed contributions of Cu²⁺ and *p* oxygen holes. This would suggest a charge transfer from the chain to the planes since no low-energy pre-peak is visible in the planes of YBCO₆.

These spatially resolved measurements thus allow us to postulate more directly the localization of the charge and to identify the differences between the two compounds, due to the multiple inequivalent oxygen and copper sites, much more directly than the polarization-dependent XAS.

Spatial resolution and comparison with polarized XAS. The spatial resolution of the O *K* edge is examined in greater detail in Fig. 3a and in Supplementary Fig. 2, in which the spectra are isolated into contributions from the O(1) (chains), O(2,3) (planes) and O(4) (apical) sites. A similarity in the shape of the O *K* edge spectra corresponding to the chains and apical positions is clearly observed, indicating that the apical sites are associated with the chains rather than the planes. However, the finite spatial resolution of EELS spectrum image (SI) (with a Gaussian broadening due to the dynamics of the electron beam propagation of around 4 Å full width at half maximum) results in some contributions to the spectra at the apical O(4) position from both chain O(1) and plane O(2,3) oxygen sites (see Supplementary Figs 5 and 6), limiting our ability to clearly distinguish such closely spaced atoms. As stated previously, the intensity of the Cu *L* edge from the chains is significantly lower than the one coming from the planes, as can be seen on Supplementary Fig. 2. This might explain the differences compared with XAS measurements (Supplementary Figs 3 and 4).

Beam exposure induced doping dependence. In addition to investigating the O₆ and O₇ end member samples, we also investigated the doping dependence of the EELS in YBCO by

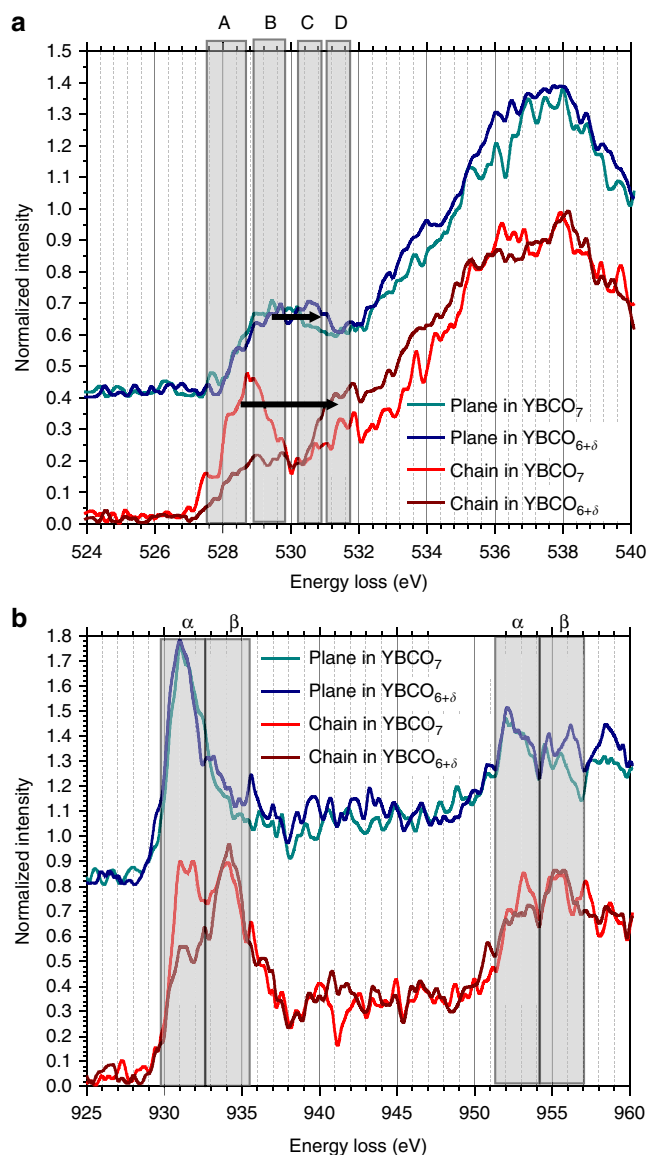


Figure 4 | Changes in the electronic structure of chains and planes

during electron beam exposure. Effect of the loss of oxygen in the chains on the fine structures of the O *K* (a) and Cu *L* (b) for planes and chains of YBCO₇ [100]. Spectra of planes have been shifted vertically for clarity. Arrows show charge transfer in planes (from the low-energy band to the UHB) and chains (from ‘full chain states’ to ‘empty chain states’) happening upon removal of chain oxygen through beam exposure.

measuring the YBCO₇ sample as a function of exposure to the electron beam, which has for effect the removal of the loosely bound O(1) intercalated interstitial oxygen from the chain layer (in contrast the O₆ sample, which has no O(1) and is relatively insensitive to beam exposure). This experiment provides a novel means to investigate the doping dependence (converting YBCO₇ to YBCO_{6+δ}) of the electronic structure in the planes and chains by examining the EELS of YBCO₇ subject to different levels of beam exposure. On the O *K* edge (Fig. 4a), the beam exposure in the chains results in a reduction in peak A, associated with full chain states, and an enhancement in peak D, associated with ‘empty’ chains. In the planes, the beam exposure results in a transfer of spectral weight from the mobile carrier band (A) to the UHB states (B). Both of these observations are consistent with the beam exposure resulting in the removal of oxygen and subsequent reduction in the doping of the material.

Similarly, on the Cu *L* edge (Fig. 4b), there is a modest change in the planes with beam exposure, which is not surprising given the relative insensitivity to doping previously seen. In the chain layer, however, peak α drops significantly and peak β increases slightly with beam exposure, consistent with the changes observed in Fig. 2c and roughly in agreement with a progressive reduction of the Cu(1) from Cu²⁺ to Cu⁺.

Discussion

Although the agreement between polarization-dependent XAS and our spatially resolved EELS is generally compelling, indicating a largely consistent understanding of the site-specific orbital occupation and symmetry in YBCO, there remain two notable discrepancies. First, from Fig. 2a (as well as Supplementary Fig. 6), the pre-peak corresponding to the hole states in the chains appears stronger at the side (close to the BaO(4) layer) than in the centre of the chain (Cu(1)O(1) atomic plane), suggesting that the holes are predominantly situated on the apical oxygen O(4). This observation runs contrary to expectations for the formation of a $3d_{z^2-y^2}$ -symmetry Zhang–Rice singlet-like state forming around the Cu(1) in the full chains. Accordingly, since the O(1) is co-ordinated by two Cu(1) atoms, whereas O(4) is co-ordinated by only one Cu(1) atom, it is expected that on a per-atom basis the hole occupation of O(1) is larger than that of O(4). This discrepancy may be associated with the shorter Cu(1)–O(4) bond length (1.833 Å) relative to the Cu(1)–O(1) bond (1.940 Å), and potentially a role of strong correlation effects altering the expected electronic structure. An additional consideration is related to the dynamics of electron scattering as the probe propagates in the sample, generating some contributions from adjacent sites, as shown recently in La₂CuO₄ by Lugg and co-workers²⁴. The second discrepancy relates to the presence of the peak β at 934.5 eV in the Cu *L* chain layer of fully doped YBCO₇. This peak has been associated with empty chain sites, and polarization-dependent XAS measurements in YBCO₇ show no evidence for a peak at this photon energy³, yet the peak is present in the EELS spectra of the fully doped O₇ sample. This presence may indicate that beam exposure in the first two unit cells has already induced some O loss. However, that this peak evolves only very weakly with further beam exposure (underdoping) presents the possibility that it may yet be intrinsic. This possibility requires confirmation with further experiments at lower temperatures, and possibly at lower voltages, at which the mobility of species and knock-on effects can be reduced.

In conclusion, by exploiting recent developments in electron microscopy, such as aberration correction, we are able to image the real-space electronic structure in the planes and chains of doped and undoped YBCO. These measurements directly show, in real space, the transfer of charge from the chain layer to the CuO₂ planes when the chains are intercalated with excess oxygen. Using EELS at the Cu *L* and O *K* edges, it is verified that the doped holes in the planes go primarily into the O 2*p* rather than the Cu 3*d* states, with an accompanying transfer of spectral weight from the UHB to a band of states near the Fermi energy as shown in La_{2-x}Sr_xCuO₄ that have only a CuO₂ plane layer^{20,21,23}. In contrast to the planes, the chain layer exhibits large changes at both the Cu *L* and O *K* edges, indicative of a change in valence from Cu¹⁺ to Cu²⁺ in the chains. Together with symmetry-resolved but spatially integrated XAS measurements, a consistent description of the doping-dependent unoccupied states of YBCO is obtained.

Methods

Sample preparation. Single crystals of YBa₂Cu₃O_{6+ δ} with stoichiometries (δ = 0.03 and 0.97) were grown and detwinned, as described elsewhere²⁵. Fresh TEM samples, in the [100] and [010] directions from the YBa₂Cu₃O_{6.97} single

crystal and one sample in the [100] using the YBa₂Cu₃O_{6.03} crystal were prepared using an NVision 40 dual-beam focused ion beam (FIB) (Carl Zeiss AG, Germany) equipped with a 30-kV Ga⁺ ion gun. The sample was thinned down to a relative thickness of $t/\lambda = 0.7$ (where t is the absolute thickness and λ is the inelastic mean free path of the electrons in the TEM at the selected accelerating voltage of 80 kV and estimated at 20 nm) using progressively decreasing ion beam energies in the FIB down to 1 keV. This process was followed by Ar⁺ milling with 900 eV ions at liquid nitrogen temperature using a Fischione Model 1040 NanoMill until a thickness of around $0.4t/\lambda$ (that is, about 20 nm according to the mean free path estimator plugin for Digital Micrograph²⁶) was reached. The localized milling area obtained using this instrument helps reduce damage of the sample during thinning while cooling at LN₂ temperature.

Experimental details. EELS measurements were carried out on a FEI Titan 80–300 Cubed (Scanning) Transmission Electron Microscope ((S)TEM) (FEI Company, Eindhoven, The Netherlands) equipped with a high-brightness X-FEG electron source, aberration correctors for probe and image lenses, a monochromator and a high-resolution electron energy loss spectrometer 866 model Gatan Imaging Filter Tridium. The experiments were performed at 80 keV in STEM mode where a fine probe of about 1 Å was rastered onto the specimen. The energy resolution was 0.7 eV, as the monochromator was not excited for this particular work.

It is important to note that, to reduce ionization and oxygen removal from the chains, a very low electron beam current (< 30 pA) was used. From each pixel, the high-angle scattered electrons were collected using a high-angle annular dark field detector while an EELS spectrum was recorded using the spectrometer.

Dispersion and exposure time of 0.5 eV/channel and 0.05 s per pixel, respectively, were used for atomic mapping of the O *K*, Ba *M* and Cu *L* edges. The O *K* and Cu *L* edges' fine structures were recorded in Spectrum Imaging mode using 0.1/channel and 0.2 eV/channel dispersions, respectively, and a collection time of 0.05 s per pixel. The spectrometer was tuned on the Ba *M*_{4,5} edges. A collection angle of 82 mrad can be achieved in energy-filtered STEM mode, allowing most of the scattered electrons in the EELS entrance aperture. XAS data were taken from previously reported measurements by Hawthorn *et al.*⁷

Data processing. The 2D projection of the EELS spectrum from Fig. 1c–e and Supplementary Figs 5c and 6c were realized, summing the spectra from every pixel on each horizontal line to reduce the statistical noise and rebinning the spectrum image in the *y* direction (*c* axis) by 2. Principal component analysis (PCA) was used for Fig. 1b and Supplementary Figs 1, 5b,d and 6b,d. A reconstruction using the first 10 principal components was performed using the weighted-PCA MSA Plugin²⁷ within Gatan Digital Micrograph software. The 2D projection of the EELS spectrum from Supplementary Figs 5d and 6d was realized summing the spectra from every pixel of the PCA-reconstructed data on each horizontal line to reduce the statistical noise. Spectra from Figs 2 and 3, and Supplementary Figs 2c–f, 4a,b and 6a,b were obtained by decomposition of the Spectrum Image (SI) in atomic planes (each with the same number of pixels) and summing equivalent atomic sites over 3 unit cells. All spectra were aligned in energy using the Ba *M*₅ peak for SI containing Ba and this alignment was matched for high dispersion data of the O *K* edge. All spectra were normalized between 0 and 1. For comparison purposes a Gaussian broadening of 0.7 eV full width at half maximum was applied to the EELS and XAS spectra from Fig. 2d and Supplementary Figs 3 and 4.

References

- Batson, P. & Chisholm, M. Anisotropy in the near-edge absorption fine structure of YBa₂Cu₃O_{7- δ} . *Phys. Rev. B* **37**, 635–637 (1988).
- Fink, J. *et al.* Electron energy-loss and x-ray absorption spectroscopy of cuprate superconductors and related compounds. *J. Electron Spectros. Relat. Phenomena* **66**, 395–452 (1994).
- Grigis, C. & Schamm, S. Element and phase identification via fine structure analysis in EELS: application to MOCVD-Y₁Ba₂Cu₃O_{7- δ} thin films. *Ultramicroscopy* **74**, 159–167 (1998).
- Nücker, N. *et al.* Symmetry of holes in high-T_c superconductors. *Phys. Rev. B* **39**, 6619–6629 (1989).
- Browning, N. D., Yuan, J. & Brown, L. M. Determination of the local oxygen stoichiometry in YBa₂Cu₃O_{7- δ} by electron energy loss spectroscopy in the scanning transmission electron microscope. *Phys. C Supercond* **202**, 12–18 (1992).
- Browning, N. D. & Pennycook, S. J. *Characterization of High T_c Materials and Devices by Electron Microscopy* (Cambridge University Press, 2010).
- Hawthorn, D. G. *et al.* Resonant elastic soft x-ray scattering in oxygen-ordered YBa₂Cu₃O_{6+ δ} . *Phys. Rev. B* **84**, 075125 (2011).
- Peets, D. C. *et al.* X-Ray absorption spectra reveal the inapplicability of the single-band Hubbard model to overdoped cuprate superconductors. *Phys. Rev. Lett.* **103**, 087402 (2009).
- Nücker, N. *et al.* Site-specific and doping-dependent electronic structure of YBa₂Cu₃O_x probed by O 1s and Cu 2*p* x-ray-absorption spectroscopy. *Phys. Rev. B* **51**, 8529–8542 (1995).

10. Krol, A. *et al.* Hole states in CuO_2 planes and Cu-O chains of $\text{YBa}_2\text{Cu}_3\text{O}_7$ and $\text{YBa}_2\text{Cu}_4\text{O}_8$ probed by soft-x-ray absorption spectroscopy. *Phys. Rev. B* **45**, 2581–2584 (1992).
11. Yarmoff, J. *et al.* Valence electronic structure of $\text{Y}_1\text{Ba}_2\text{Cu}_3\text{O}_7$. *Phys. Rev. B* **36**, 3967–3970 (1987).
12. Kuiper, P. *et al.* X-ray absorption study of the O 2p hole concentration dependence on O stoichiometry in $\text{YBa}_2\text{Cu}_3\text{O}_x$. *Phys. Rev. B* **38**, 6483–6489 (1988).
13. Botton, G. A., Lazar, S. & Dwyer, C. Elemental mapping at the atomic scale using low accelerating voltages. *Ultramicroscopy* **110**, 926–934 (2010).
14. Lazar, S. *et al.* Imaging, core-loss, and low-loss electron-energy-loss spectroscopy mapping in aberration-corrected STEM. *Microsc. Microanal.* **16**, 416–424 (2010).
15. Bosman, M. *et al.* Two-dimensional mapping of chemical information at atomic resolution. *Phys. Rev. Lett.* **99**, 086102 (2007).
16. Egerton, R. F. Limits to the spatial, energy and momentum resolution of electron energy-loss spectroscopy. *Ultramicroscopy* **107**, 575–586 (2007).
17. Chien, T. Y. *et al.* Visualizing short-range charge transfer at the interfaces between ferromagnetic and superconducting oxides. *Nat. Commun.* **4**, 2336 (2013).
18. Turner, S. *et al.* Atomic resolution coordination mapping in $\text{Ca}_2\text{FeCoO}_5$ brownmillerite by spatially resolved electron energy-loss spectroscopy. *Chem. Mater.* **24**, 1904–1909 (2012).
19. Grioni, M. Unoccupied electronic structure and core-hole effects in the x-ray-absorption spectra of Cu_2O . *Phys. Rev. B* **45**, 3309–3318 (1992).
20. Chen, C. *et al.* Out-of-plane orbital characters of intrinsic and doped holes in $\text{La}_{2-x}\text{Sr}_x\text{CuO}_4$. *Phys. Rev. Lett.* **68**, 2543–2546 (1992).
21. Romberg, H., Alexander, M., Nücker, N., Adelman, P. & Fink, J. Electronic structure of the system $\text{La}_{2-x}\text{Sr}_x\text{CuO}_{4+\delta}$. *Phys. Rev. B* **42**, 8768–8771 (1990).
22. Liang, R., Bonn, D. A. & Hardy, W. N. Evaluation of CuO_2 plane hole doping in $\text{YBa}_2\text{Cu}_3\text{O}_{6+x}$ single crystals. *Phys. Rev. B* **73**, 180505 (2006).
23. Chen, C. *et al.* Electronic states in $\text{La}_{2-x}\text{Sr}_x\text{CuO}_{4+\delta}$ probed by soft-x-ray absorption. *Phys. Rev. Lett.* **66**, 104–107 (1991).
24. Lugg, N. R. *et al.* Removing the effects of elastic and thermal scattering from electron energy-loss spectroscopic data. *Appl. Phys. Lett.* **101**, 183112 (2012).
25. Liang, R., Bonn, D. A. & Hardy, W. N. Growth of YBCO single crystals by the self-flux technique. *Philos. Mag.* **92**, 2563–2581 (2012).
26. Malis, T., Cheng, S. C. & Egerton, R. F. EELS log-ratio technique for specimen-thickness measurement in the TEM. *J. Electron Microsc. Tech.* **8**, 193–200 (1988).
27. Bosman, M., Watanabe, M., Alexander, D. T. L. & Keast, V. J. Mapping chemical and bonding information using multivariate analysis of electron energy-loss spectrum images. *Ultramicroscopy* **106**, 1024–1032 (2006).

Acknowledgements

We are thankful to Julia Huang for FIB TEM sample preparation. We gratefully acknowledge the financial support from NSERC (Discovery Grant) to G.A.B. The experimental microscopy work was carried out at the Canadian Centre for Electron Microscopy, a National Facility supported by NSERC and McMaster University.

Author contributions

G.A.B. and N.G. conceived the experiments. N.G. performed the STEM-EELS experiments and data processing, and prepared the draft manuscript. D.G.H. provided the anisotropic XAS data. W.N.H., R.X.L. and D.A.B. prepared the samples (crystal growth, annealing, de-twinning). G.A.B. supervised the experiments and the data analysis. D.G.H. and G.A.S. contributed to the manuscript preparation. All authors discussed the results and implications and commented on the manuscript at all stages.

Additional information

Supplementary Information accompanies this paper at <http://www.nature.com/naturecommunications>

Competing financial interests: The authors declare no competing financial interests.

Reprints and permission information is available online at <http://npg.nature.com/reprintsandpermissions/>

How to cite this article: Gauquelin, N. *et al.* Atomic scale real-space mapping of holes in $\text{YBa}_2\text{Cu}_3\text{O}_{6+\delta}$. *Nat. Commun.* 5:4275 doi: 10.1038/ncomms5275 (2014).