

## CO<sub>2</sub> Sensors

Ocean Optics CO<sub>2</sub> Sensors provide a quick and easy way to monitor levels of Dissolved Carbon Dioxide in a variety of environments. While the importance of medical and biochemical applications dominate the CO<sub>2</sub> detection landscape, recent years have revealed a growing demand for environmental monitoring of CO<sub>2</sub>, particularly sweet/sea water analysis. Ocean optics fiber optic CO<sub>2</sub> sensor features a wide dynamic range and a high sensitivity required for medical, biochemical, and environmental applications.

### About CO<sub>2</sub> Sensors

The following sections contain information about Ocean Optics' Fiber Optic CO<sub>2</sub> Sensors:

#### Principle of Fiber Optic CO<sub>2</sub> Sensors

Ocean Optics' CO<sub>2</sub> sensor is based on a proprietary sol-gel film (composed of Aluminum silicate co-polymer) that is coated on the tip of optical fibers. We immobilize the sol-gel film with a highly stable pH sensitive fluorescence molecule (8-hydroxypyrene-1,3,6-trisulfonic acid, HPTS). We then protect the sol-gel film with a hydrophobic silicone coating that is permeable to CO<sub>2</sub> and impermeable to H<sup>+</sup>. The immobilized sol-gel film creates a pH sensitive solid-state (semi-aqueous) microenvironment at the tip of the fiber.

The sol-gel formulation and process is designed to keep the HPTS indicator in its deprotonated form in the absence of CO<sub>2</sub>. The ambient CO<sub>2</sub> diffuses through the hydrophobic silicone coating into the sol-gel film and change the equilibrium between the HPTS protonated and deprotonated forms. An Ocean Optics miniature spectrometer monitors the fluorescence of the deprotonated form (at 515 nm maximum, excited at 470 nm), and the changes in fluorescence intensity are correlated with the CO<sub>2</sub> partial pressure.

#### Advantages of Sol-gel Technology

Sol-gel is an inorganic glass with high thermal and thermomechanical stability and optical transparency compared to organic polymers. These features makes the Ocean Optics sol-gel based CO<sub>2</sub> sensor more stable at high temperatures compared to CO<sub>2</sub> sensors using polymers for encapsulation of HPTS. Additionally, flexibility when immobilizing high concentrations of indicators as well as the presence of multiple indicators is much higher than polymers. Lastly, glasses such as sol-gel have a higher photochemical stability than polymers.

#### CO<sub>2</sub> Sensor System Components

##### Hardware

The CO<sub>2</sub> sensor hardware is similar to FOXY, with the exception of the 450 nm shortpass filter in front of the blue LED. Most FOXY probe designs can also be used for CO<sub>2</sub> sensor. However, the sol-gel coating on these probes must be changed.

##### Software

The software is going to be part of OOISensors. It will be similar to FOXY, with some minor modifications.

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## Using the CO<sub>2</sub> Sensor System

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The following sections provide information on handling and using Ocean Optics Fiber Optic CO<sub>2</sub> Sensors.

### Handling CO<sub>2</sub> Sensors

Avoid bringing the CO<sub>2</sub> sensor into contact with hard objects. The CO<sub>2</sub> sensor has a layer of soft, black silicone coated on the outside of its tip that can be damaged by excess pressure or contact with hard objects.

Precondition the CO<sub>2</sub> sensor before use. Submerge and suspend the tip of the sensor in water for approximately two hours. Ensure that the tip of the sensor is not in contact with any other substance except for the water during this preconditioning to prevent damage to the silicone tip. All CO<sub>2</sub> sensors must be preconditioned prior to use.

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**Note:** After each use, submerge and suspend the tip of the CO<sub>2</sub> sensor in water. By keeping the sensor in a preconditioned state, you can use it whenever it is needed. If you take the sensor out of water after use, you must precondition the sensor prior to using it again.

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### Using CO<sub>2</sub> Sensors

Follow the steps below to use the CO<sub>2</sub> Sensors:

#### Configuring a Single Point Calibration

1. Open the CO<sub>2</sub> Sensors software.
2. Select **Calibrate | Oxygen/CO<sub>2</sub>, Single Temperature** from the menu. The Single Temperature Calibration window appears.
3. Select **Multi Point** as the calibration type, if necessary, from the Calibration Type options located on the top of the screen.
4. Insert the Calibration floppy disk that came with your precalibrated CO<sub>2</sub> sensor into the floppy disk drive.
5. Select **File | Open Calibration Table**, and then select the calibration table stored on the floppy disk.
6. Click the **Curve Fit** button located on the right side of the window.
7. Click the **Update Channel Calibration** button.

#### Performing a Single Point Calibration

1. Select **Single Point** from the Calibration Type box on the screen.
2. Place the CO<sub>2</sub> sensor in a solution with a known CO<sub>2</sub> concentration that falls within the precalibrated concentration range.
3. Enter the known concentration in the Concentration box.
4. Select the **Continuous** check box on the right side of the screen, if necessary.
5. Press the **Scan Standard** button on the right side of the screen.
6. Allow the intensity signal on the Intensity box to stabilize. This will take approximately 10 minutes.
7. Press the **Scan Standard** button again to stop the scan.
8. Press the **Curve Fit** button.
9. Press the **Update Channel Calibration** button and click the OK button when the pop-up window appears.
10. Select **File | Close** to exit the Single Temperature Calibration window.

## CO<sub>2</sub> Sensors

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### Beginning a Scan

1. Select the **Cont.** option on the Scan switch located at the bottom right of the screen.
2. Click the **Scan** switch up to begin scanning.

### Storing CO<sub>2</sub> Sensors

When not using your CO<sub>2</sub> sensors, store the sensor probe submerged in water. The probe tip should be submerged in water for at least two days before testing.

In order to preserve the life of the sensor, you should turn the excitation light off when not using the sensor to avoid photobleaching.

## CO<sub>2</sub> Standard Solutions

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The following equations comprise the CO<sub>2</sub> standard solutions:

Henry's coefficient,  $H = 10^{1.464} = 29.107$  [(atm\*L)/mol] at 25 ° and 1 atm pressure.

### How to Make CO<sub>2</sub> Standard Solutions:

If we make solutions of known NaCO<sub>3</sub> concentrations and keep the solution at low pH (~3) then all the CO<sub>2</sub> is dissolved. The CO<sub>2</sub> partial pressure can be calculated using Henry's Coefficient.

@ 25 °C, 1 atm

$H = \text{CO}_2 \text{ fugacity [atm]} / \text{CO}_2 \text{ concentration [mole/L]} = 10^{1.464}$  [(atm\*L)/mole]

$X \text{ mol\%} = (\text{partial pressure} / \text{total pressure}) * 100$

### Example:

@ 25 °C, 1 atm

$10^{-4}$  [mol/L] in solution □ 0.29 [mol %] in gas □ 0.0029 atm partial pressure

$10^{-3}$  [mol/L] in solution □ 2.9 [mol %] in gas □ 0.029 atm partial pressure

$2 \times 10^{-3}$  [mol/L] in solution □ 5.8 [mol %] in gas □ 0.058 atm partial pressure

$4 \times 10^{-3}$  [mol/L] in solution □ 11.6 [mol %] in gas □ 0.116 atm partial pressure

### Pre-calibrated CO<sub>2</sub> Range

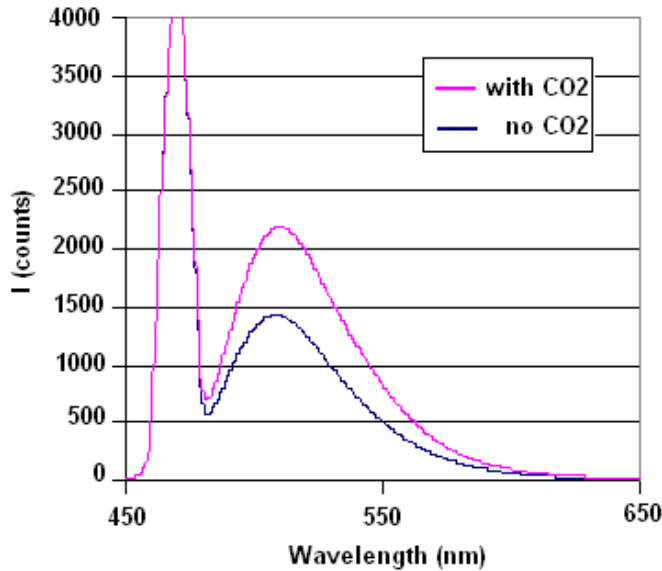
Mole % CO <sub>2</sub>	Partial Pressure (atm)
0	0
4.4	0.044
6.5	0.065
9.6	0.096
12.9	0.129

## CO<sub>2</sub> Sensor Specifications

The following charts illustrate the various responses of the CO<sub>2</sub> sensor in differing situations:

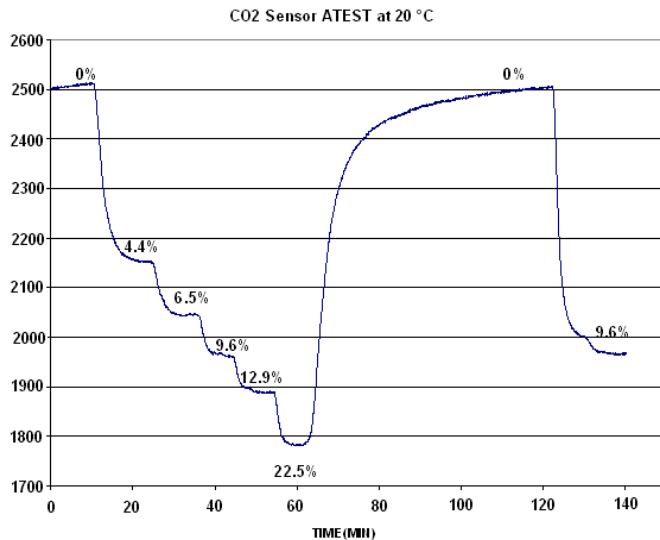
### Sensor Spectra

The following figure illustrates the sensor spectrum, which shows the back reflection of excitation light at 470 nm and fluorescence of HPTS at 515 nm. This graphic clearly shows how CO<sub>2</sub> quenches the fluorescence intensity.



### Dynamic Response

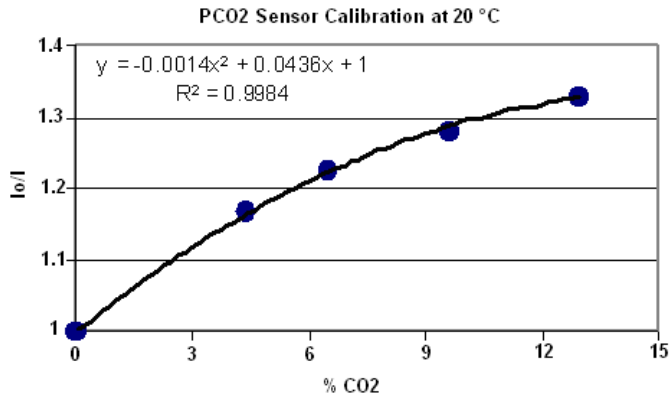
The graphic below illustrates the dynamic response of the CO<sub>2</sub> sensor (CO<sub>2</sub> % V.S.I.). It illustrates the reproducibility as well as the reversibility of the sensor. The sensor is sensitive to CO<sub>2</sub> from 0% to 100%, with higher sensitivity readings at lower concentrations. The practical range of CO<sub>2</sub> detection is from 0-25%.



## CO<sub>2</sub> Sensors

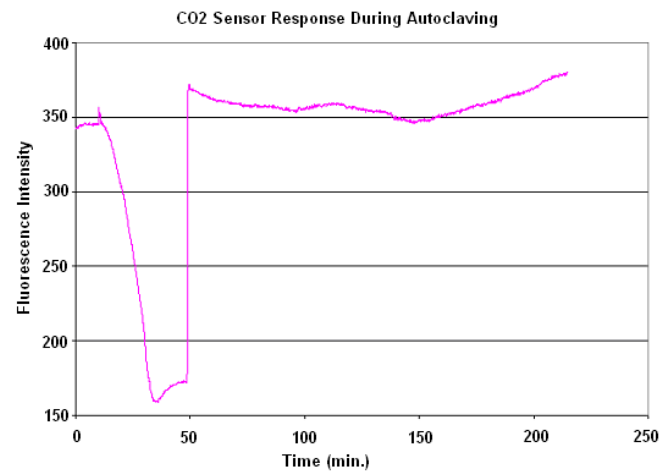
### Calibration

The figure below illustrates sensor calibration between 0 and 12.5% CO<sub>2</sub>. The calibration data fits a second-order polynomial relation between the inverse of fluorescence intensity and CO<sub>2</sub> concentrations.



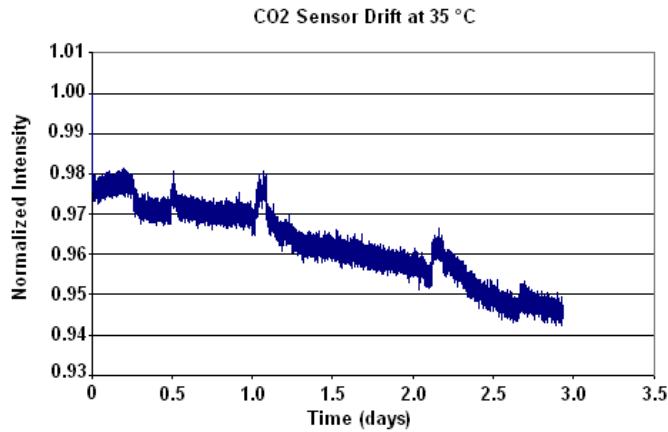
### Sensor Sterilization

Our CO<sub>2</sub> fiber optic sensor can be sterilized at 121 C and 15 PSI as shown in the following Figure. This figure shows the sensor fluorescence time line during sterilization for about half hour. The graph shows the full recovery of fluorescence after sterilization.



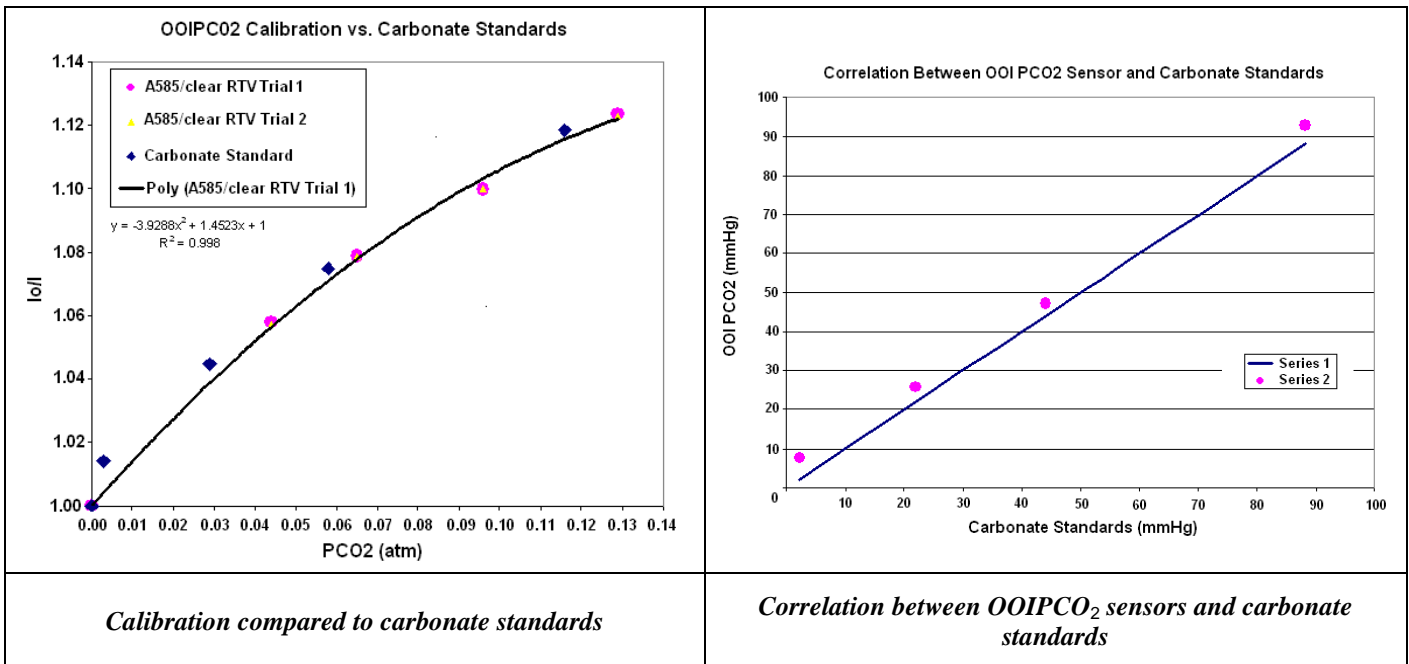
**Sensor drift**

Sensor signal drift is primarily caused by the photobleaching of the HPTS. The sensor drift was about 5% of actual intensity reading in 3 days which corresponds to 0.01% CO<sub>2</sub> (absolute reading of CO<sub>2</sub> concentration). The Figure below shows the sensor drift in 3 days. You can minimize the signal drift by reducing the LED intensity. Maximum LED intensity (20 mA) was used in this experiment. Previous experiments with oxygen sensors (FOXY) indicate that a reduction of the LED current to 5 mA reduces the photobleaching effectively.



**Carbonate Standards Comparison**

The following graphics illustrate the correlations between Ocean Optics CO<sub>2</sub> Sensors and carbonate standards:



## General Specifications

Measurement	PCO <sub>2</sub>
Range	1-100% at atmospheric pressure (760 mm Hg) 0-25% practical range
Resolution	0.003% (30 μatm) at low concentrations 0.03% (300 μatm) at high concentrations
Lowest detectable limit	0.003% (30 μatm)
Response time	Average of 10 minutes for large CO <sub>2</sub> changes Less than 10 minutes for less than 1% changes (under 3% CO <sub>2</sub> ) <b>Note:</b> Delay in response time is due to the thick silicone coating on the sensor. Using thinner coatings can reduce response time.
Sensor lifetime	To be determined
Sterilization	Autoclavable (121 °C, 15 PSI)
Operating temperature	To be determined (currently tested up to 120 °C)
Calibration	Single point and Multi-point calibration (contact an Application Scientist for calibration procedures). The preferred reference is a CO <sub>2</sub> -free water (for example, a water saturated with N <sub>2</sub> ).
Temperature calibration	To be determined
Probe information	1/16 " OD probe (1000 μm core diameter) in a stainless steel ferule with SMA termination
Disposable probes	Disposable sensor tips will be available in future. The disposable CO <sub>2</sub> tips can be attached to a reflective probe
Recommended spectrometer configuration	SF2000 or USB2000 with the USB-LS-450-C Blue LED Pulsed Light Source and a new band pass filter
Storage	Store in water for a few hours before using.